Palladium-Catalyzed Asymmetric Alkylation of 2,3-Alkadienyl Phosphates. Synthesis of Optically Active 2-(2,3-Alkadienyl)malonates

Yasushi Imada,* Katsuya Ueno, Koji Kutsuwa, and Shun-Ichi Murahashi*

Department of Chemistry, Graduate School of Engineering Science, Osaka University, 1-3 Machikaneyama, Toyonaka, Osaka 560-8531 Department of Chemistry, Faculty of Engineering Okayama University of Science, 1-1 Ridaicho, Okayama 700-0005

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Asymmetric alkylation of 2,3-alkadienyl phosphates with soft carbon nucleophiles proceeds efficiently in the presence of palladium complex catalyst bearing MeOBIPHEP or BINAP ligand to give optically active functionalized allenes up to 90% ee.

Optically active allenes bearing axial chirality are important synthetic intermediates for various naturally occurring biologically active compounds.¹ Therefore, much attention has been paid to the asymmetric synthesis of this class of compounds. Optically active allenes have been prepared by either optical resolution of racemic allenes¹ or chirality transfer reactions of optically active precursors.²;³ Preparation of optically active allenes using chiral catalyst is an attractive method; however, the reported methods give moderate enantioselectivity (20–60% ee)⁴ except kinetic resolution by asymmetric oxidative degradation of allenes.⁵ Recently, Hayashi et al. reported that palladium-catalyzed reaction of 2-bromo-1,3 butadienes gives optically active allenes with enantioselectivity up to 89% ee.⁶ We wish to report that palladium-catalyzed asymmetric reaction of racemic 2,3-alkadienyl phosphates 1 with soft carbon nucleophiles 2 gives optically active functionalized allenes 3 (Scheme 1).⁷ This is based on the non-asymmetric reaction reported by Gore et al.⁸

Alkylations of 2,3-alkadienyl phosphates 1 with soft carbon nucleophiles 2 were examined in the presence of N , O -bis(trimethylsilyl)acetamide (BSA) and the palladium catalyst prepared from $Pd_2(dba)_3$ •CHCl₃ and MeOBIPHEP⁹ or BINAP¹⁰ in THF at room temperature (Scheme 1). The representative results of the reactions are summarized in Table 1.

Table 1. Asymmetric alkylation of diethyl 2,3-alkadienyl phosphates 1 with malonates 2^a

Entry		Phosphate Malonate Time/h Product Yield/% ^b				$Ee/\%$ ^c
	1a	2a	8	3a	76	73
	1 _b	2a	6	3 _b	72	77
3	1c	2a	6	3c	89	81
	1d	2a	12	3d	69	90
	1e	2a	8	3e	80	60
6	1a	2 _b		3f	85	76
	1a	2c		3g	74	69
	1f	2d	8	3 _h	90	70

 $^{\text{a}}$ All reactions were carried out in THF (0.1 M) at room temperature. The ratio of 1/2/BSA/Pd2(dba)3-CHCl3/(R)-MeOBIPHEP was 100/110/120/ 1/4. ^bIsolated yield based on 1. ^cDetermined by HPLC analysis with chiral stationary phase column.

The palladium-catalyzed reactions of phosphates 1a–e with diethyl 2-acetamidomalonate (2a) give the corresponding functionalized allenes 3a–e with 60–90% ee in good yields (entries 1–5). The catalytic reactions of the phosphate 1a with diethyl 2 phenylmalonate (2b) and diethyl 2-methylmalonate (2c) gave 3f and 3g, respectively, with similar enantio-selectivities (entries 6 and 7), indicating that the acetamido group is not essential for getting high level of enantioselectivity. Noteworthy is that the reaction of the phosphate 1f with dimethyl malonate (2d) afforded monosubstituted malonate 3h as a sole product. Disubstituted malonate could not be detected among the products (entry 8).

Typically, asymmetric reaction of diethyl 5,5-dimethyl-2,3 hexadienyl phosphate (1d) with acetamidomalonate 2a was carried out in THF at room temperature for 12 h in the presence of BSA and the palladium catalyst prepared in situ from $Pd_2(dba)_3$ [.]CHCl₃ (0.01) equiv) and (R)-MeOBIPHEP (0.04 equiv). Ethyl 2-acetamido-2 ethoxycarbonyl-7,7-dimethyl-4,5-octadienoate $[(-)$ -3d] was obtained with 90% ee in 69% isolated yield. The absolute configuration of $(-)$ -3d was determined to be (R) on the basis of the Lowe's rule.¹¹

A survey of chiral bidentate phosphine ligands revealed that the C_2 ligands bearing binaphthyl and biphenyl backbone, such as MeOBIPHEP and BINAP are effective.

The palladium-catalyzed reaction of 2,3-alkadienyl phosphates 1 with soft carbon nucleophiles⁸ can be rationalized by assuming the mechanism shown in Scheme 2. The first step is facile formation of 1-3- η ³-1,3-alkadienylpalladium 4 from allyl esters bearing highly reactive leaving group $OP(O)(OEt)₂$.¹² Soft carbon nucleophiles attack at the less hindered α -position of the palladium complex 4. It is in contrast to the ν -attack of hard carbon nucleophiles, $8,13$ such as organozinc and Grignard reagents, and carbon monoxide¹⁴ via η ¹-1,3-alkadien-3-yl-palladium 5 to give conjugated dienes.

An equilibrium is present between two diastereomeric η^3 alkadienylpalladium complexes (S_p) -4 and (R_p) -4 through η^1 alkadienylpalladium complex 5. Preferential nucleophilic attack of malonate anion to (S_p) -4 at the less hindered allylic terminus from Scheme 1. the opposite side to the palladium would give the optically active

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functionalized allene (R) -3.

Fast equilibrium between two diastereomeric η^3 -alkadienylpalladium complexes (S_p) -4 and (R_p) -4 was confirmed by the following control experiment. When enantiomerically pure (S) -1a¹⁵ was allowed to react with malonate 2a in the presence of BSA and the palladium catalyst, prepared from $Pd_2(dba)$ ₃ \cdot CHCl₃ and 1,4bisdiphenylphosphinobutane, in THF at room temperature for 2 h, the racemic allene 3a was obtained in 76% yield.

The participation of η ¹-alkadienylpalladium complex 5 was confirmed from the reaction of phosphate 1b with Pd(PPh₃)₄. The ¹H and ³¹P NMR spectra of the product derived from the reaction of the phosphate 1b with an equivalent of $Pd(PPh_3)_4$ in C_6D_6 in an NMR tube showed a facile, exclusive formation of η^1 -pentadienylpalladium 5a.¹⁶ The chloro complex 5b was obtained by ligand exchange reaction upon treatment with zinc chloride as shown in Scheme 3. The structure of 5b was confirmed by single-crystal Xray analysis.¹⁷

In summary, we succeeded in asymmetric synthesis of functionalized allenes 3 with high enantioselectivity by the palladium-catalyzed asymmetric alkylation of racemic 2,3-alkadienyl phosphates 1 using palladium–MeOBIPHEP catalyst. Further mechanistic study and application are in progress.

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References and Notes

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- 15 Enantiomerically pure (S)-1a { $[\alpha]_D$ 48.1° (c 1.1, CHCl₃)} was obtained by the optical resolution of racemic 1a by HPLC with a chiral stationary phase (Chiralcel OD).
- 16 5a: ¹H NMR (500 MHz, C₆D₆, selected data) δ 0.60 (t, J = 7 Hz, 6 H), 1.57 $(d, J = 7 \text{ Hz}, 3 \text{ H}, \text{H}^e), 3.25 (dq, J = 7, 7 \text{ Hz}, 4 \text{ H}), 4.51 (br, 1 \text{ H}, \text{H}^d), 4.60 (d,$ $J = 10$ Hz, H^b), 5,27 (dd, $J = 10$, 17 Hz, H^c), 6,13 (d, $J = 17$ Hz, H^a). 5b: ¹H NMR (500 MHz, C_6D_6 , selected data) δ 1.72 (d, $J = 7$ Hz, 3 H, H^e), 4.71 (dd, $J = 2$, 11 Hz, H^b), 4.97 (br, 1 H, H^d), 5,68 (ddd, $J = 2$, 10, 16 Hz, H^c), 6,07 (dd, $J = 2$, 16 Hz, H^{a}), 31 P{¹H} NMR (202 MHz, $\mathrm{C}_6\mathrm{D}_6$) δ 25.73. For related $(\eta^1$ -1,3-butadien-2-yl)palladium complexes generated from Pd(PPh₃)₄ and 2,3-butadienyl chlorides, see: S. A. Benyunes, L. Brandt, A. Fries, M. Green, M. F. Mahon, and T. M. T. Papworth, J. Chem. Soc., Dalton Trans., 1993, 3785.
- 17 Crystal data for $5b \cdot 3C_6H_6$: C₅₉H₅₅ClP₂Pd, fw = 967.82, Rigaku RAXIS-RAPID, monoclinic, Cc (No. 9), $a = 20.1149(10)$ A, $b = 17.3353(11)$ A, $c = 13.9252(8)$ A, $\beta = 94.843(2)$ °, $V = 4838.4(5)$ A³ $Z = 4$, $T =$ 101(2) K, $D_{\text{calcd}} = 1.329 \text{ Mg m}^{-3}$, $\lambda = 0.71069 \text{ A}$, $F(000) = 2008$, $R_1 =$ 0.0314; $wR_2 = 0.0676$ for all 5532 reflections, GOF = 1.040.